Acta Cryst. (1978). B34, 164–167

# The Crystal and Molecular Structure of Methyl Tetraacetyl-β-D-glucoside

By P. ZUGENMAIER AND G. RAPPENECKER

Institut für Makromolekulare Chemie der Universität Freiburg, Stefan-Meierstrasse 31, D-7800 Freiburg, Federal Republic of Germany

(Received 23 March 1977; accepted 15 June 1977)

The crystal structure of methyl 2,3,4,6-tetra-O-acetyl- $\beta$ -D-glucopyranoside has been determined from 2260 intensity data collected on a four-circle diffractometer with Mo K $\alpha$  radiation. The unit cell is orthorhombic, space group  $P2_12_12_1$ , a = 8.089, b = 13.251, c = 17.359 Å, and contains four molecules. The structure was solved with direct methods (*MULTAN*). Full-matrix, weighted, least-squares refinement gave a final agreement index of R = 0.043 for 1577 observed reflections.

#### Introduction

The progress in solving complex polymer structures in recent years is mostly due to the introduction of advanced methods, such as conformation and packing analysis, into this field (Zugenmaier & Sarko, 1972, 1973, 1976; Arnott & Scott, 1972; Guss *et al.*, 1975). Precise molecular data of various model compounds are necessary for the application of these methods. A survey for polysaccharides of such data as valence bond lengths, angles, and torsion angles is given by Arnott & Scott (1972). However, very few data are

available for models of polysaccharide derivatives. Some acetyl derivatives have been solved in recent years. In continuing the investigation of the acetyl compounds, we have studied the crystal and molecular structure of methyl tetraacetyl- $\beta$ -D-glucoside and compared the results with those of  $\beta$ -D-acetylcellobiose (Leung, Chanzy, Pérez & Marchessault, 1976).

### Crystal data

 $C_{15}O_{10}H_{22}, M_r = 362.3$ , orthorhombic,  $P_{2,2,2,1}, a = 8.089 \pm 0.003, b = 13.251 \pm 0.005, c = 17.359 + 12.251 \pm 0.005, c = 17.359$ 

## Table 1. Fractional atomic coordinates ( $\times 10^4$ ; for H $\times 10^3$ ) for methyl tetraacetyl- $\beta$ -D-glucoside

The estimated standard deviations are given in parentheses and refer to the last decimal positions of respective values.

	X	y	Ζ		x	ŗ	Ζ
O(1)	4092 (3)	3420 (2)	5230 (2)	H(1)	210 (6)	262 (4)	494 (3)
O(2)	2916 (3)	3551 (2)	3651 (2)	H(2)	490 (6)	202(1) 271(3)	304 (3)
O(3)	3563 (3)	1785 (2)	2800 (2)	H(3)	180 (6)	186(4)	350 (3)
O(4)	2528 (4)	4 (2)	3643 (2)	H(4)	471 (6)	63 (3)	386 (3)
O(5)	4074 (3)	1713 (2)	5127(2)	H(5)	202 (6)	85 (3)	403 (3)
O(6)	5749 (4)	-144(2)	5056 (2)	H(6A)	346 (6)	-70(4)	508 (3)
O(2A)	5340 (5)	4386 (3)	3556 (3)	H(6R)	390 (7)	-1(4)	584 (3)
O(3A)	940 (5)	1745 (5)	2366 (3)	H(11)	364 (7)	283 (4)	632 (4)
O(4A)	4807 (5)	-909 (3)	3362 (3)	H(12)	422 (7)	409 (4)	627 (4)
O(6A)	5690 (6)	-1788(3)	5296 (3)	H(13)	234(7)	378 (4)	605(4)
C(1)	3347 (4)	2626 (3)	4845(2)	H(21)	171(8)	514(5)	321 (4)
C(2)	3721 (4)	2690 (3)	3992 (2)	H(22)	320 (0)	570 (5)	321(4)
C(3)	3001 (4)	1777(3)	3586 (2)	H(23)	278 (8)	J 19 (J) A05 (5)	312(4)
C(4)	3545 (4)	799 (3)	3966 (2)	H(21)	276 (8)	495 (5)	230 (4)
C(5)	3287 (5)	832 (3)	4831 (3)	H(32)	220 (8)	127(4)	155 (3)
C(6)	4004 (5)	-70(3)	5245 (3)	H(32)	229 (6)	190 (4)	111 (3)
C(1M)	3482 (5)	3539 (4)	6006 (3)	H(33) H(41)	400 (8)	239 (4)	147(3)
C(2A)	3872 (6)	4338 (3)	3/33 (3)	$\Pi(41)$ $\Pi(42)$	149(11)	122 (0)	261 (5)
C(3A)	2383 (6)	1788 (4)	2228 (2)	H(42)	201(11)	-202(6)	270 (5)
C(AA)	2303(0)	1700 (4) 816 (2)	2230 (3)	H(43)	144 (11)	-1/8(6)	331 (5)
C(4A)	5517 (6) 6444 (6)	-610(3)	5005 (3)	H(01)	8/5(8)	-50(5)	511 (4)
C(0A)	2951 (9)	-1039(3)	3093 (3)	H(62)	869 (8)	-163(5)	503 (4)
C(2M)	2031(0)	$\frac{5111}{4}$	3021 (3)	H(63)	830(8)	-105(5)	426 (4)
C(SM)	3180(7)	1830 (4)	1465 (3)				
C(4M)	2096 (10)	-154/(4)	3016 (4)				
U(0M)	8211(7)	-1034(3)	4845 (4)				



Fig. 1. Projection of one molecule of methyl tetraacetyl- $\beta$ -D-glucoside on the xy plane after application of the rotation R to the coordinates of Table 1.

0.006 Å, V = 1860.4 Å<sup>3</sup>,  $D_c = 1.293$ ,  $D_x = 1.297 \pm 0.007$  g cm<sup>-3</sup>, Z = 4, F(000) = 768, Mo  $K\alpha$  ( $\lambda = 0.7107$  Å).

Suitable crystals were obtained by recrystallizing methyl tetraacetyl- $\beta$ -D-glucoside from saturated acetone solution allowed to evaporate slowly. Precession and Weissenberg photographs exhibited *mmm* Laue symmetry with alternating extinction along the axes indicating the orthorhombic space group  $P2_12_12_1$ .

A crystal of dimensions  $0.6 \times 0.8 \times 1.2$  mm was mounted on a glass fiber with **a** along the spindle axis for data collection. Intensity data were collected at room temperature with a fully automated Nonius fourcircle diffractometer. The intensities of two standard reflections were measured periodically during data collection. 2260 independent reflections were recorded of which 1577 had intensities greater than  $2\sigma_i$ ;  $\sigma_i$  being the standard deviation in intensities which was estimated from the total count and the background values. The intensity data were corrected for Lorentz, polarization and absorption effects. In the least-squares



Fig. 2. Projection of methyl tetraacetyl- $\beta$ -D-glucoside molecules of the crystal on the *bc* plane. Only molecules contributing to close contacts are shown for the illustration of packing.

refinement, extinction corrections were applied to the Table 2. Bond lengths (Å), bond angles and torsion weighted structure factors.

The phasing model was obtained with MULTAN (Main, Woolfson & Germain, 1971). In the refinement cycles the C and O atoms were refined anisotropically. The H atoms were assigned the isotropic U values of the atoms to which they were bonded covalently and only the coordinates were refined. The scattering factors were taken from International Tables for X-ray Crystallography (1962). The final R factor obtained with 1577 reflections with  $\sigma_1 > 2$  is 4.3%, the R value of  $F^2$  is 4.6%. The XRAY 70 and 76 systems (Stewart, 1970/76) were used throughout the calculations.

#### **Results and discussion**

The final coordinates of methyl tetraacetyl- $\beta$ -Dglucoside are listed in Table 1.\* Fig. 1 shows the numbering of the atoms and gives a projection of one molecule on the xy plane after application of the rotation R to the coordinates of Table 1.

$$\mathsf{R} = \begin{pmatrix} 0.419 & 0.680 & 0.603\\ 0.851 & -0.524 & 0.000\\ 0.316 & 0.513 & -0.798 \end{pmatrix}.$$

Valence bond lengths, bond angles and torsion angles are listed in Table 2. In Fig. 2 the packing of the molecules in the crystal is illustrated and Table 3 shows the shortest packing contacts.

The pyranose ring of methyl tetraacetyl- $\beta$ -Dglucoside (TMG) is in the C1 chair conformation and O(6) is placed in the vicinity of the  $gg^{\dagger}$  position. The ring conformation of TMG compares best with the unprimed ring of  $\beta$ -D-acetylcellobiose (AC) (Leung et al., 1976) while the chair conformation of the primed ring of AC is more pronounced and O(6) is placed near the gt position. Only the C(5)-O(5)-C(1)-C(2) ring torsion angle of  $-67.7^{\circ}$  of TMG exceeds considerably the average value of  $-62 \cdot 2^{\circ}$  (Arnott & Scott, 1972) as do the angles for both AC rings, -70.1 and  $-69.8^{\circ}$ . The O(5)-C(1)-O(1)-C(1M) torsion angle of  $-76.7^{\circ}$  for the O-C(methyl) bond for TMG agrees well with the O(5)–C(1)–O(1)–C(4') angle of  $-77 \cdot 1^{\circ}$ in AC, connecting two residues. The torsion angles for bonds to pendant ring atoms of TMG lie within 6° of those for AC.

The valence bond lengths of the ring are in agreement with the average values (Arnott & Scott, 1972). Only the C-O lengths to pendant O atoms of

# angles (°) for methyl tetraacetyl- $\beta$ -D-glucoside

The estimated standard deviations are given in parentheses and refer to the last decimal positions of respective values.

Ring			
$\begin{array}{c} C(1) & C(2) \\ C(2) - C(3) \\ C(3) + C(4) \end{array}$	1·514 (6) 1·516 (5) 1·520 (5)	$O(5) \cdot C(1) - C(2)$ C(1) - C(2) - C(3) $C(2) - C(3) \cdot C(4)$	107.5 (3) 109.5 (3) 111.6 (3)
C(4) - C(5)	1.516 (6)	C(3) - C(4) - C(5)	111.4 (3)
C(5) O(5)	1.427 (5)	C(4) - C(5) - O(5)	108.6(3)
O(5) = C(1)	1.431 (5)	C(3) = O(3) = C(1)	112.0 (3)
Pendant atoms			
C(1)O(1)	1.385 (5)	O(5)-C(1)-O(1)	107.3 (3)
C(2) -O(2)	1.441 (4)	C(2)-C(1)-O(1) C(1)-C(2) O(2) C(3)-C(2)-O(2)	$ \begin{array}{c} 110.1(3) \\ 110.8(3) \\ 105.5(3) \end{array} $
C(3)-O(3)	1.439 (5)	C(2) - C(3) - O(3)	108.3 (3)
C(4) -O(4)	1.449 (5)	C(4)-C(3)-O(3) C(3)-C(4)-O(4) C(5)-C(4)-O(4)	$   \begin{array}{r}     109.1 (3) \\     106.7 (3) \\     109.0 (3)   \end{array} $
C(5)C(6)	1.510 (6)	C(4) - C(5) - C(6)	113.4 (3)
		O(5)-C(5)-C(6)	107.7 (3)
C(6) - O(6)	1.453 (5)	C(5)-C(6)-O(6)	108.6 (3)
Methyl groups			
O(1) - C(1M)	1.443 (5)	C(1)-O(1)-C(1M)	112.6 (3)
Acetyl groups			
O(2)-C(2A)	1.352 (5)	C(2) - O(2) - C(2A)	117.9 (3)
C(2A)-O(2A)	1.209 (6)	O(2)-C(2A)-O(2A)	123.6 (4)
C(2A)-C(2M)	1.497 (7)	O(2) - C(2A) - C(2M)	110.2(4)
O(3) - C(3A)	1.365 (6)	C(3) = O(3) = C(3A)	(4) 120.2 (4) 117.3 (3)
C(3A) O(3A)	1.190 (7)	O(3)-C(3A)-O(3A)	123.5 (4)
C(3A) - C(3M)	1.491 (7)	O(3) - C(3A) - C(3M)	110.0 (4)
	1 2(4 (()	O(3A) - C(3A) - C(3A)	4) $126.5(5)$
O(4) - C(4A) C(4A) - O(4A)	1.304(0) 1.212(8)	C(4) = C(4) = C(4A) O(4) = C(4A) = O(4A)	117.5 (4)
C(4A) = O(4A) C(4A) = C(4M)	1.495 (9)	O(4) - C(4A) - C(4M)	110.7(5)
		O(4A)-C(4A)-C(4A)	<i>A</i> ) 127.0 (5)
O(6) - C(6A)	1.339 (5)	C(6) - O(6) - C(6A)	117.2(3)
C(6A) = O(6A) C(6A) = C(6M)	1.494 (0)	O(6) - C(6A) - O(6A) O(6) - C(6A) - C(6A)	$122\cdot 2(3)$
	1 171 (0)	O(6A) - C(6A) - C(6A)	<i>M</i> ) 126-3 (4)
Torsion angles Ring			
O(5)–C(	(1) - C(2) - C(3)	3) 59.5	(3)
C(1)-C	(2)-C(3)-C(4)	4) -52.3	(4)
C(2)-C	(3) - C(4) - C(4)	$\frac{5}{2}$ 49.3	(4)
C(3) = C(3)	(4) - C(5) - O(3)	$-53 \cdot 1$ (1) $-64 \cdot 1$	(4)
C(5)-O	(5) - C(1) - C(2)	(2) -67.7	(3)
-(-, 0)			
Pendant	atoms		
C(5)-O	(5)-C(1)-O(	1) 173.9	(3)
O(5)-C	(1) - C(2) - O(2)	2) 175-4	(2)
C(1)-C(2)	(2) - C(3) - O(3)	$-172 \cdot 3$	(3)
C(2) - C( C(3)=C(	(3)-C(4)-O(4) (4)-C(5)-C(4)	+) 168·2 5)172.9	(3)
O(5) C	(5) - C(6) - O(6)	$-64 \cdot 1$	(4)

<sup>\*</sup> Lists of structure factors and thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 32942 (19 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

<sup>+</sup> gg means: the bond C(6)-O(6) is gauche to O(5)-C(5) and gauche to C(4)-C(5).

$\begin{array}{c} C(3)-C(2)-C(1)-O(1)\\ C(4)-C(3)-C(2)-O(2)\\ C(5)-C(4)-C(3)-O(3)\\ O(5)-C(5)-C(4)-O(4)\\ C(1)-O(5)-C(5)-C(6)\\ C(4)-C(5)-C(6)-O(6) \end{array}$	176.0 (3) -171.5 (3) 168.9 (3) -170.6 (3) -172.7 (3) 56.1 (4)
Methyl group	
O(5)-C(1)-O(1)-C(1 <i>M</i> )	-76.7 (3)
Acetyl groups	
$\begin{array}{c} C(1)-C(2)-O(2)-C(2A)\\ C(2)-O(2)-C(2A)-O(2A)\\ C(2)-O(2)-C(2A)-C(2M)\\ C(2)-C(3)-O(3)-C(3A)\\ C(3)-O(3)-C(3A)-O(3A)\\ C(3)-O(3)-C(3A)-C(3M)\\ C(3)-C(4)-O(4)-C(4A)\\ C(4)-O(4)-C(4A)-O(4A)\\ C(4)-O(4)-C(4A)-C(4M)\\ \end{array}$	$108.7 (4) \\ -5.2 (6) \\ 174.2 (3) \\ -122.6 (3) \\ -3.0 (7) \\ 176.7 (3) \\ 127.5 (4) \\ 2.4 (6) \\ -178.4 (4)$
C(5)-C(6)-O(6)-C(6A) C(6)-O(6)-C(6A)-O(6A)	-153.8(3) -2.0(6)
C(6)-O(6)-C(6A)-C(6M)	176.9 (4)

the ring are slightly longer for TMG, with the exception of C(1)–O(1).

Remarkable are the very small C(3)-C(2)-O(2)and C(5)-C(6)-O(6) bond angles of  $105 \cdot 5$  and  $108 \cdot 6^{\circ}$ , respectively, which deviate from the established average values by  $5 \cdot 3$  and  $3 \cdot 2^{\circ}$ , but have corresponding values in either the primed or unprimed residue of AC.

The bond lengths, angles, and torsion angles of the four acetyl groups of TMG vary little. The acetyl groups lie in a plane with the adjacent O and C atoms. The *cis* positions of O(2A), O(3A) and O(4A) relative to the corresponding ring H atoms are clearly revealed in Fig. 1.

This work was supported by a grant from Deutsche Forschungsgemeinschaft. The authors thank Dr W. Littke for data collection on the Nonius diffractometer

Table 3. Short packing contacts (Å)

$H(4) \cdots H(13^{i})$	2.27	$O(3A) \cdots C(4M^{iii})$	3.405
$H(12)\cdots H(3^{i})$	2.46	$C(1M) \cdots O(3A^{i})$	3.476
$O(4A) \cdots H(33^{ii})$	2.47	$C(4A)\cdots O(2A^{ii})$	3.482
$O(1) \cdots H(5^i)$	2.57	$C(3M) \cdots O(2A^{ii})$	3.485
$O(5) \cdots H(1^i)$	2.60	$O(4A) \cdots O(2A^{ii})$	3.355
$C(1M) \cdots H(3^i)$	2.83	$O(6) \cdots O(2^i)$	3.545
$C(1) \cdots H(1^{i})$	3.07	$O(6) \cdots O(1^i)$	3.574
$O(4A)\cdots C(2A^{ii})$	3.311	$C(6M)\cdots C(1M^i)$	3.629
$O(4A)\cdots C(2M^{ii})$	3.344	$C(6)\cdots C(3M^{iv})$	3.634
$O(4A)\cdots C(3M^{iii})$	3.393	$C(6A)\cdots C(2M^i)$	3.683
Symmetry code			

(i)	$x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$	(ii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$
(iii)	$-x, y + \frac{1}{2}, -z + \frac{1}{2}$	(iv) $-x + \frac{1}{2}, -y, z + \frac{1}{2}$

and Dr U. Drück for valuable discussions. The computations were carried out at the Computer Center of the University of Freiburg.

## References

- ARNOTT, S. & SCOTT, W. E. (1972). J. Chem. Soc. Perkin II, pp. 324–335.
- GUSS, J. M., HUKINS, D. W. L., SMITH, P. J. C., WINTER, W. T., ARNOTT, S., MOORHOUSE, R. & REES, D. A. (1975). J. Mol. Biol. 95, 359–384.
- International Tables for X-ray Crystallography (1962). Vol. III, p. 202. Birmingham: Kynoch Press.
- LEUNG, F., CHANZY, H. D., PÉREZ, S. & MARCHESSAULT, R. H. (1976). Can. J. Chem. 54, 1365-1371.
- MAIN, P., WOOLFSON, M. M. & GERMAIN, G. (1971). MULTAN. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univ. of York, England.
- STEWART, J. M. (1970/76). The XRAY system. Tech. Rep. TR-446. Computer Science Center, Univ. of Maryland, College Park, Maryland.
- ZUGENMAIER, P. & SARKO, A. (1972). Acta Cryst. B28, 3158–3166.
- ZUGENMAIER, P. & SARKO, A. (1973). Biopolymers, 12, 435–444.
- ZUGENMAIER, P. & SARKO, A. (1976). *Biopolymers*, 15, 2121–2136.